BINDURA UNIVERSITY OF SCIENCE EDUCATION CHEMISTRY DEPARTMENT

MASTER OF SCIENCE DEGREE IN CHEMISTRY

COURSE: MCH503: ORGANIC CHEMISTRY

3 HOURS

AUG 20124

ANSWER ANY FOUR QUESTIONS. EACH QUESTION CARRIES 25 MARKS

- 1. a) What do you understand by the phrase enantiomeric resolution? [2 marks]
 - b) 1-phenylethylamine has the structure shown below.

Describe how you would resolve a racemic mixture of 1-phenylethylamine.

[10 marks]

c) Workout names of the compounds below.

$$H_2C$$
 O CH_3 CH_3 CH_3 CH_3 CH_3 CH_3

[4 marks]

- d) Discuss tannins under the following headings;
 - (i) Structure

[2 marks]

(ii) Detection methods

[5 marks]

(iii) Application in education

[2 marks]

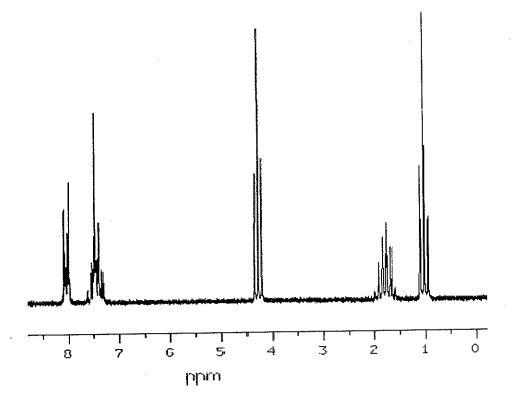
2. a) Predict the normal C-13, DEPT-90, and DEPT-135 spectra of a molecule, whose structure appears below. [10 marks]

$$H_2C$$
 H_3C CH_3

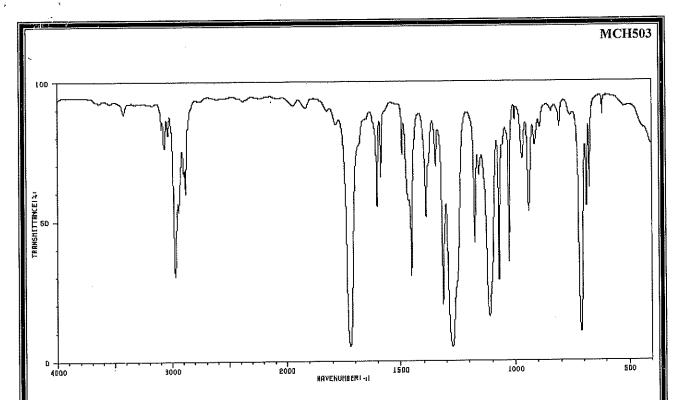
- b) Propose mechanisms to account for the formation of fragments corresponding to m/z 74, 71, 59 and 43 in the mass spectrum of methyl butyrate (C₅H₁₀O₂). [5 marks]
- c) Explain why tetramethylsilane (TMS) is used as reference compound in both ¹H- and ¹³C-NMR spectroscopy. [5 Marks]
- d) Predict the chemical shifts in ¹³C NMR spectra of the compound below; H₂C=CHCO₂CH₂CH₃ [5 marks]
- 3. Determine the structure of the compound C₁₀H₁₂O₂, given the information below;

¹³C NMR (δ, ppm): 11, 22, 67, 128, 130, 131, 133, 167

¹H NMR:



IR:



[25 marks]

4. a) Mechanistically account for the biosynthesis of shikimic acid from phosphoenolpyruvate (PEP) and D-erythrose-4-P (E4P).

[10 marks]

shikimic acid

b) Illustrate details of the conversion of acetyl CoA to isopentenyl pyrophosphate. [10 marks]

c) Why are phenolic acids important in our diet?

[5 marks]

5. a) Warfarin is a Coumarin that can be used as rat poison. It is synthesized in the laboratory using the reaction below. Draw a mechanism for the reaction.

[15 marks]

b) Suggest possible applications of the compounds shown below in industry.

[2 x 5 marks]

END OF PAPER