BINDURA UNIVERSITY OF SCIENCE EDUCATION FACULTY OF SCIENCE AND ENGINEERING DEPARTMENT OF CHEMISTRY

HBSc Chemical Technology: Part 1.1

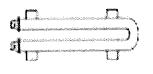
€ OCT 2024 Course Code: CH115 Industrial Chemistry I

Duration: 2 Hours

ANSWER QUESTION ONE AND ANY TWO FROM SECTION A AND ANY TWO FROM SECTION B. EACH QUESTION CARRIES TWENTY MARKS.

a) What are chemical industrial processes? Give two examples. [5 Marks] Q1

b) The following are common chemical process symbols. What is the meaning of these process symbols?



i.

ii.



[6 Marks]

- c) Write a chemical reaction equation to illustrate the production of [4 Marks] superphosphate.
- d) Draw a well-labeled process flow diagram for the industrial production of [5 Marks] calcium carbide.

SECTION A: ANSWER ANY TWO QUESTIONS FROM THIS SECTION

- a) What are industrial gases? Give any three examples other than those already Q2 [5 Marks] mentioned.
 - b) With the help of a process flow diagram, explain the industrial production of oxygen using the pressure swing adsorption (PSA) technology. [8 Marks]
 - c) Explain the production of hydrogen chloride, starting with brine water as the raw [7 Marks] material.
- a) Why are some minerals mined and others not? Q3

[4 Marks]

•	b) c) d)	What are the three steps involved in the extraction of metals? Explain the effect of reactivity on the method of metal extraction. What is the chemical structure of the following mineral ores? i. Haematite ii. Magnetite iii. Bauxite iv. Pentlandite	[6 Marks] [6 Marks] [4 Marks]
Q4	a)b)c)	Explain the pyrometallurgical approach for the extraction of nicked With the help of appropriate reaction describe the Mond Process for production of high purity nickel. What are the advantages and disadvantage of nickel over other medical series of the extraction of nickel over other medical series.	for the
Q5	SE a) b)	Give examples of any two major copper minerals. Explain with the help of reaction equations where appropriate, the steps for the extraction of copper from copper pyrites. i. Ore concentration ii. Roasting of concentrate iii. Smelting process iv. Bessemerisation.	[2 Marks]
Q6	a) b) c)	Describe the drive process for the manufacture of Portland ceme	[3 Marks] nt. [9 marks]
Q7	a)b)c)	Explain the causes and effects of eutrophication.	[4 Marks] [6 Marks] [4 Marks]

d) Explain with the help of appropriate reaction equations, the removal of phosphorus from effluent water. [6 Marks]

END OF EXAM